Impact of peatland restoration on water treatability

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Introduction

- Current UK catchment management practices have attempted to maintain or enhance the ability of peat to immobilize organic carbon (OC).
- As such, research focus has tended towards variability of the OC flux within runoff rather than OC size and chemical character.
- These characteristics may control the lability of the OC as well as the ease at which the water treatment process can make the runoff potable.
- This study will examine the physical and chemical composition of runoff derived from the opposite members of current management programmes; moss-covered peat and bare peat (Worrall *et al.* 2004).
- The influence of runoff OC on water treatability will be assessed by examining how the type, composition and concentration of the runoff OC load affects the flocculation of $Fe(OH)_3$.



Figure 1. Two subcatchments of Crowden Great Brook, South Pennines, UK representing two possible end members of peatland catchment management; A: Bare peat, B: Vegetated peat.

Methods

- Synthetic runoff generated by passing de-ionised water through columns of intact bare peat and moss material collected from Crowden Great Brook field site.
- Physical and chemical character of runoff OC determined through TOC analysis and Py-GC/MS.
- Fe³⁺, commonly used as a flocculant in drinking • water treatment (Judd & Hillis, 2001), was added to synthetic runoff solutions made up with different types, sizes and concentrations of OC.
- Rate and degree of $Fe(OH)_3$ flocculation measured via analysis of Fe present in <1µm and <0.2µm solution filtrate using ICP-AES.

Particle Size Distribution (PSD) of Runoff OC

- PSD Great Brook peatland catchment.
- peat.
- situ streamwaters.

Chemical Composition of Runoff OC

- Aliquots of synthetic bare peat runoff and moss runoff analysed via Py-GC/MS (Agilent 5890 GC/MS).
- chromatogram identified and classified according the classification scheme presented by Vancampenhout et al. (2009).
- Runoff from bare peat characterised abundance of by compounds.
- Runoff OC from dominated bv compounds.

Effect of OC size, concentration and type on Fe(OH)₃ flocculation

- (Gaffney et al. 2008).
- increases with reducing OC concentration.
- to as low as 5 mg/L (OC).
- than moss-derived OC.
- flocculates as >1µm particles.



of OC within synthetic runoff analysed and compared with in situ data collected from the Crowden

Moss-derived runoff demonstrated to exhibit a significantly smaller PSD than runoff derived from bare

Moss-derived runoff also has a PSD more similar to in



□>1µm □<1µm, >0.2µm □<0.2µm

runoff, alongside Crowden streamwater and peat porewater samples. Standard error bars

Figure 2. Proportions of average OC (% mass of total OC) within the synthetic bare peat and moss

peaks

phenolitic

vegetated peat polysaccharide





Figure 3. Total ion chromatogram of the bare peat and vegetation peat runoff OC load. Major chromatogram peaks are labelled in accordance with the classification scheme of Vancampenhout et al. (2009). A shows a sample of the synthetic runoff derived from bare peat, **B** shows a sample of the synthetic runoff produced from moss material.

Solutions of runoff OC were made up to 10ppm Fe³⁺. Concentrations of OC and Fe³⁺ used in simulations reflective of Crowden Great Brook in situ concentrations

For both moss- and bare peat- derived OC the proportion of $Fe(OH)_3$ flocculated in >1µm particles

Flocculation of $Fe(OH)_3$ was inhibited from 20 mg/L (OC)

At lower OC concentrations (<20 mg/L) bare peatderived OC proves to be a more efficient anti-flocculant

For both runoff types, increasing the relative abundance of <0.2 μ m OC reduces the proportion of Fe(OH)₃ that



Figure 4. Average proportions of Fe (% mass of total Fe) in laboratory simulations at the point at which H⁺ production has stopped. N=3. Samples are labelled with OC concentration and OC size.

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Discussion

- The data presented here suggests that current catchment management practices aimed at to minimise restoring vegetation in order greenhouse gas production may have the additional benefit of making catchment waters easier to treat with Fe³⁺
- Runoff from bare peat proves more efficient at inhibiting flocculation of Fe complexes; this may be due to a relative abundance of phenolic groups (Garcia-Mina, 2006).
- This suggests a shift towards vegetated peat could generate more easily-treatable runoff within eroding peat catchments.
- However, the synthetic runoff produced in this study suggests a transition towards vegetated peat may increase the proportion of <0.2µm OC in catchment streams. This may offset any possible benefit associated with changes in the runoff OC chemical character.
- Further research is therefore needed into the magnitude and composition of runoff from in situ areas of vegetated peat.

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References

Gaffney, J. White, K.N. & Boult, S. (2008), Oxidation state and size of Fe controlled by organic matter in natural waters, Environmental Science and Technology 42, 3575-3581. Garcia-Mina, J.M. (2006), 'Stability, solubility and maximum metal binding capacity in metal-humic complexes involving humic substances extracted from peat and organic compost, Organic Geochemistry **37**, 1960-1972. Judd, S.J. & Hillis, P. (2001), Optimisation of combined coagulation and microfiltration for water treatment, Water Research 35, 2895-2904 Vancampenhout, K., Wouters, K., De Vos, B., Buurman, P., Swennen, R., & Deckers, J. (2009), Differences in chemical composition of soil organic matter in natural ecosystems from different climatic regions - A pyrolysis-GC/MS study, Soil Biology and Biochemistry **41**, 568-579 Worrall, F., Harriman, R., Evans, C.D., Watts, C., Adamson, J., Neal, C., Tipping, E., Burt, T.P., Grieve, I., Montieth, D., Naden, P.S., Nisbet, T., Reynolds, B., & Stevens, P., (2004), Trends in dissolved organic carbon in UK

rivers and lakes, *Biogeochemistry* **70**, 369–402.