Understanding the greenhouse effect

1) Radiation emission

2) Visible and infra-red radiation

3) The thermal equilibrium

4) Thermal equilibrium of a black enlightened surface

5) The greenhouse effect

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1) Radiation emission



Any material (here a plate, thermally insulated on one side) emit radiation and therefore loose energy



The highest the temperature is, the highest the energy emission is.

2) Visible and infra-red radiation

- a) When the temperature is higher then approximately 700°C, we can see part of the radiation emitted by the object:
- This is the visible radiation







Incandescent lamp : T= 2500 °C



Volcano Iava: T= 1000° C b) When the temperature is lower then approximately 700°C, we can't see any radiation emitted by the object:

This is the infra-red radiation

2) Visible and infra-red radiation



Normalized spectra of the radiation emitted by a material at 6000°C (the sun, black line), at 2200°C (an incandescent lamp, red line), at 700°C (volcano lava, green line) and at 30°C (blue line).



insulator

When a material receives more energy than it loses, its temperature increases.



Because its temperature increases, the energy lost by emission of radiation also increases.







... until a thermal equilibrium is reached



When a material receives less energy than it loses, its temperature decreases.



Because its temperature decreases, the energy lost by emission of radiation also decreases.







... until a new equilibrium is reached

4) Thermal equilibrium of a black enlightened surface



We consider a black plate, receiving sun light on one sight and insulated on the other sight.

Its temperature is T=0K, there is no thermal emission.



The plate receives more energy than it loses. Its temperature increases, as well as the emitted radiation.







... until a thermal equilibrium is reached.



A glass, at T=0K, is put above the black surface. The glass is transparent to the solar radiation and is opaque to the infra-red radiation.



The glass absorb the infra-red radiation emitted by the surface, it gains energy and therefore its temperature increases.



The temperature of the glass increases, it emits more radiation, half above and half below the glass.





... until a new equilibrium is reached.



The radiation emitted below the glass is absorbed by the surface, which temperature increases.



The surface temperature increases, and it emits more infra-red radiation.







... until a new equilibrium is reached.



This infra-red radiation emitted by the surface is absorbed by the glass, which temperature increases.



The temperature of the glass increases, it emits more radiation, half above and half below the glass.



... until a new equilibrium is reached.



... and so on.



... and so on.



... and so on.



If we do the sum...



If we do the sum...



If we summarize:



If we summarize: the infra-red radiation absorbed by the glass increases its temperature, it emits more radiation which is absorbed by the surface, which temperature increases.