

# Ocean acidification and its impacts on marine organisms and ecosystems

*James C. Orr*



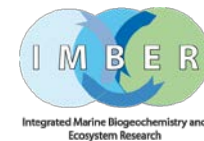
LABORATOIRE DES SCIENCES DU CLIMAT & DE L'ENVIRONNEMENT



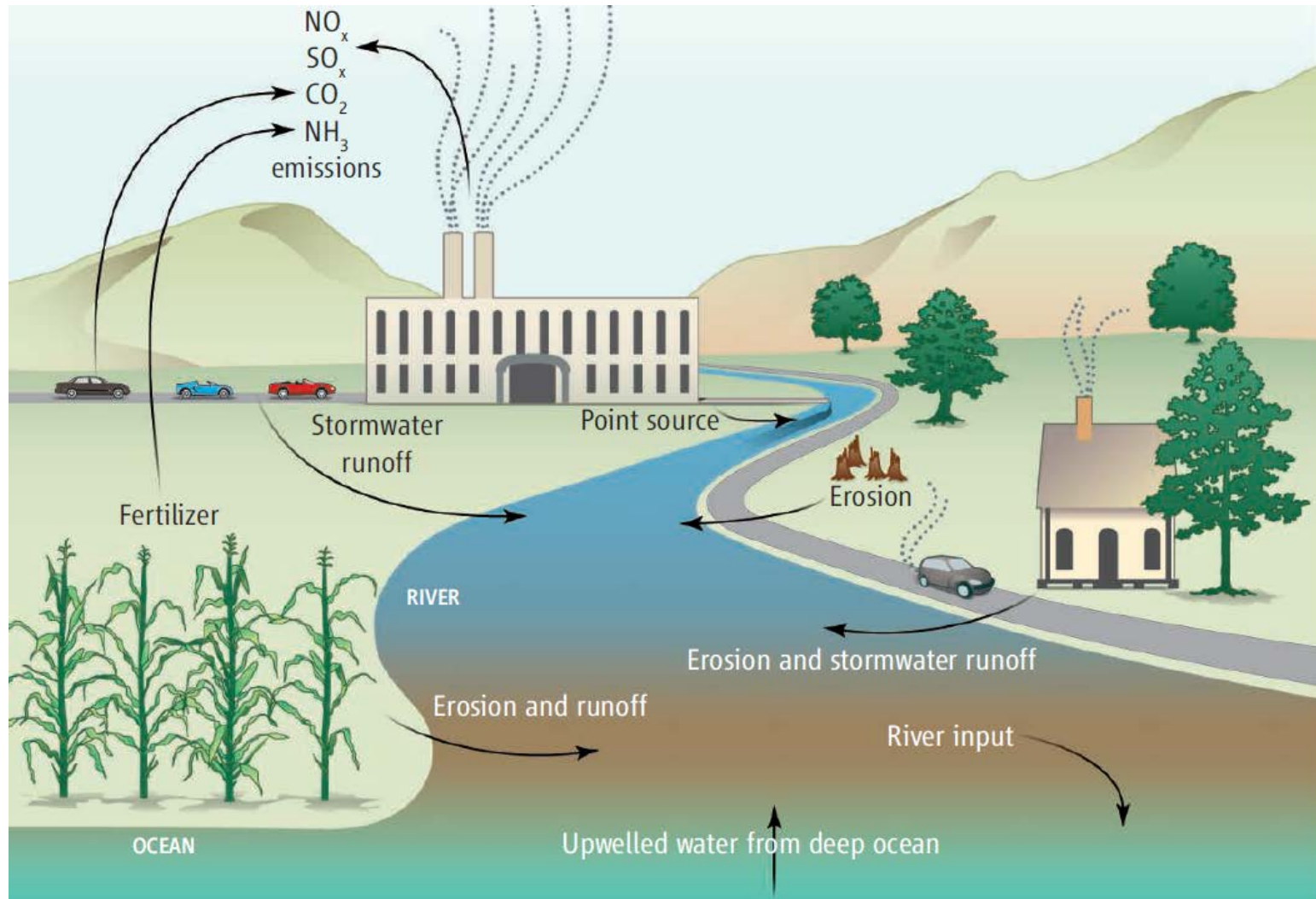
Ocean Acidification  
International  
Coordination Centre

OA-ICC

**SOLAS-IMBER Ocean Acidification Working Group**

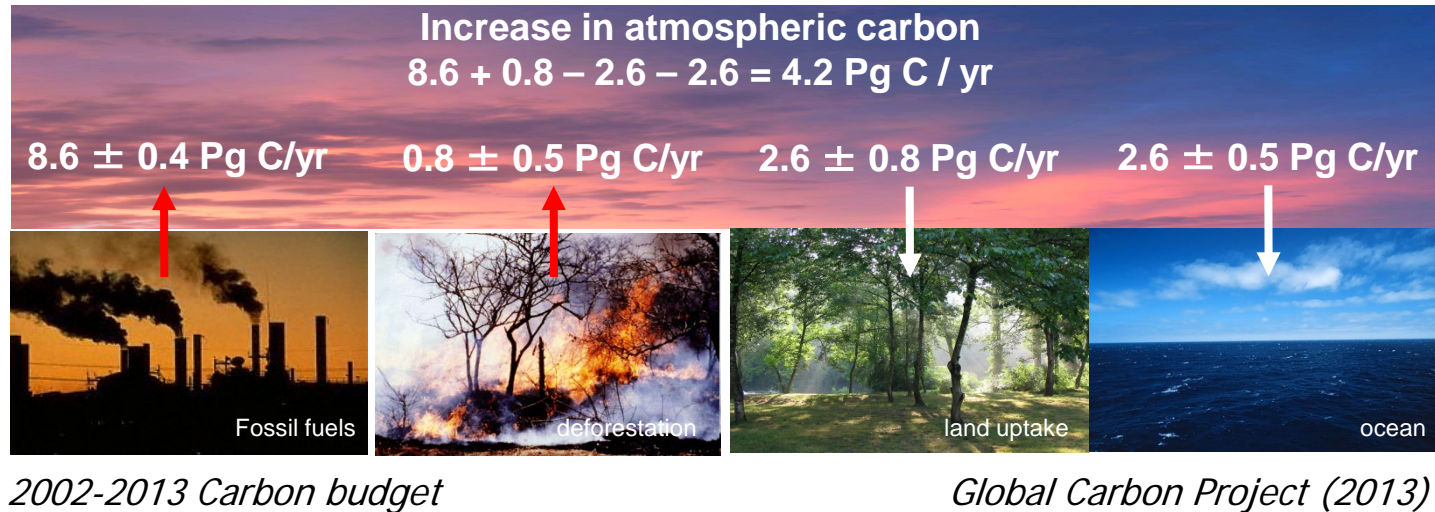


# Ocean acidification largely due to atmospheric CO<sub>2</sub> increase



Kelly et al (2011, Science)

# Ocean absorbs 1/4 of man-made CO<sub>2</sub> emissions



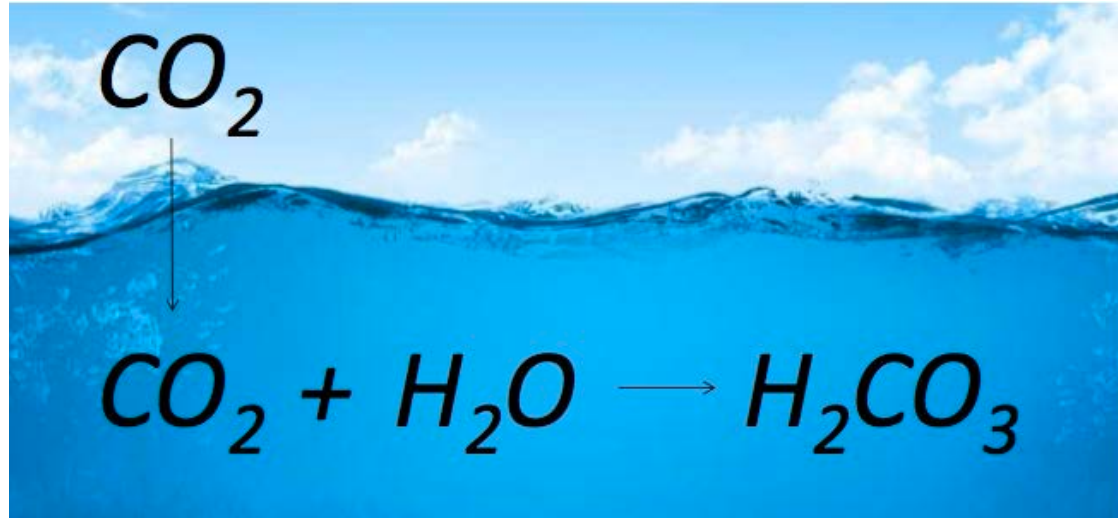
Half of emitted CO<sub>2</sub> remains in atmosphere (causing global warming)

Half absorbed by ocean & land (trees, plants, and soils)

Ocean absorbs 24 million tons of CO<sub>2</sub> every day

# More atmospheric CO<sub>2</sub> means increased ocean acidity

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*Schematic: Sam Dupont, University of Gothenburg*

CO<sub>2</sub> is an acid gas (it produces acid when combined with water)

Each of us adds 4 kg CO<sub>2</sub> per day to the ocean (increasing acidity, reducing pH)

Ocean acidity up by **26% since start of industrial age** (most in last 40 years)

Acidity could increase by **170% by 2100**

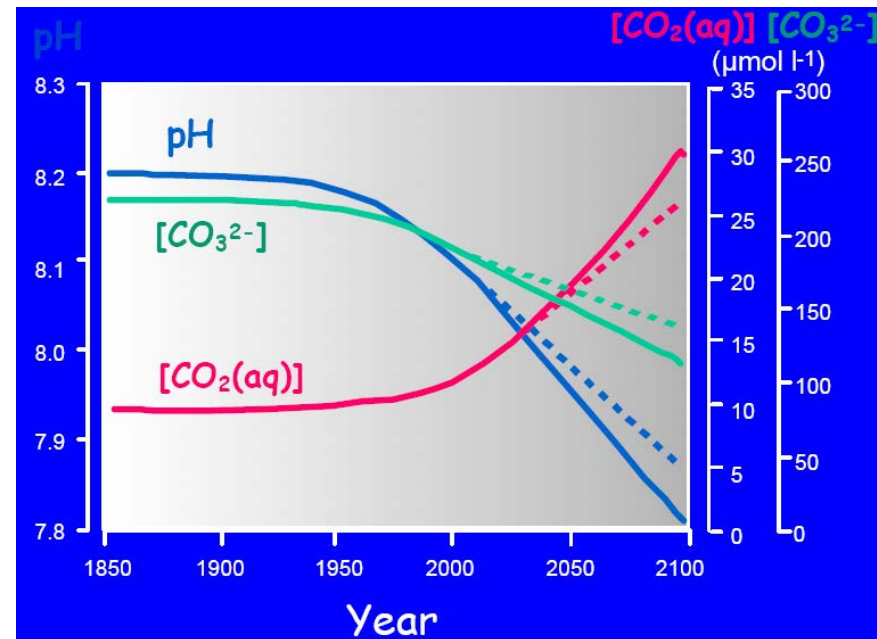
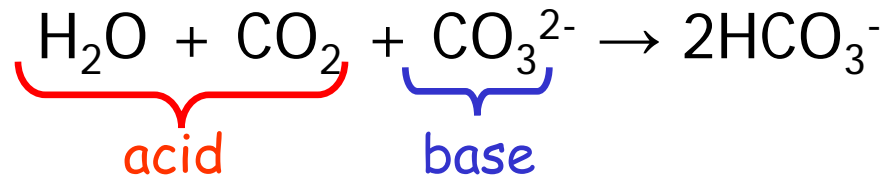
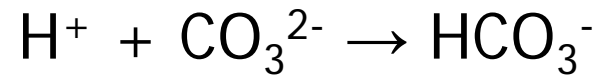
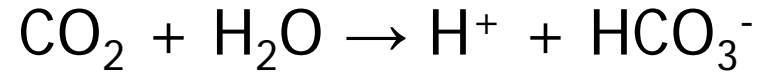
# Ocean acidification follows atmospheric CO<sub>2</sub> increase

Ocean absorbs ¼ of anthropogenic CO<sub>2</sub> emissions

CO<sub>2</sub> is an acid gas

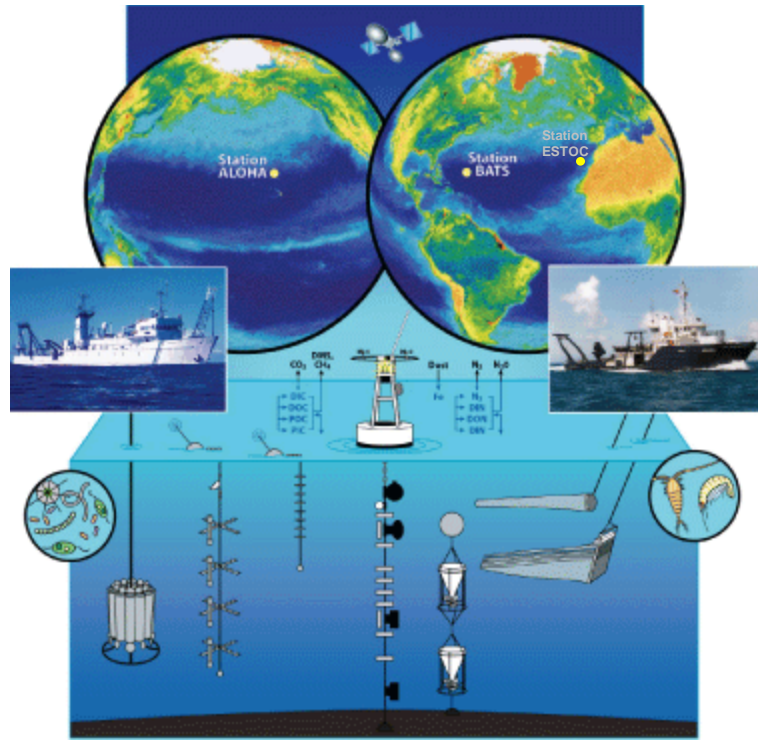
Ocean acidity +26% during industrial era, so far

$$\uparrow [H^+] = \sqrt{\frac{K'_1 K'_2 [CO_2]}{[CO_3^{2-}]}} \quad \updownarrow$$

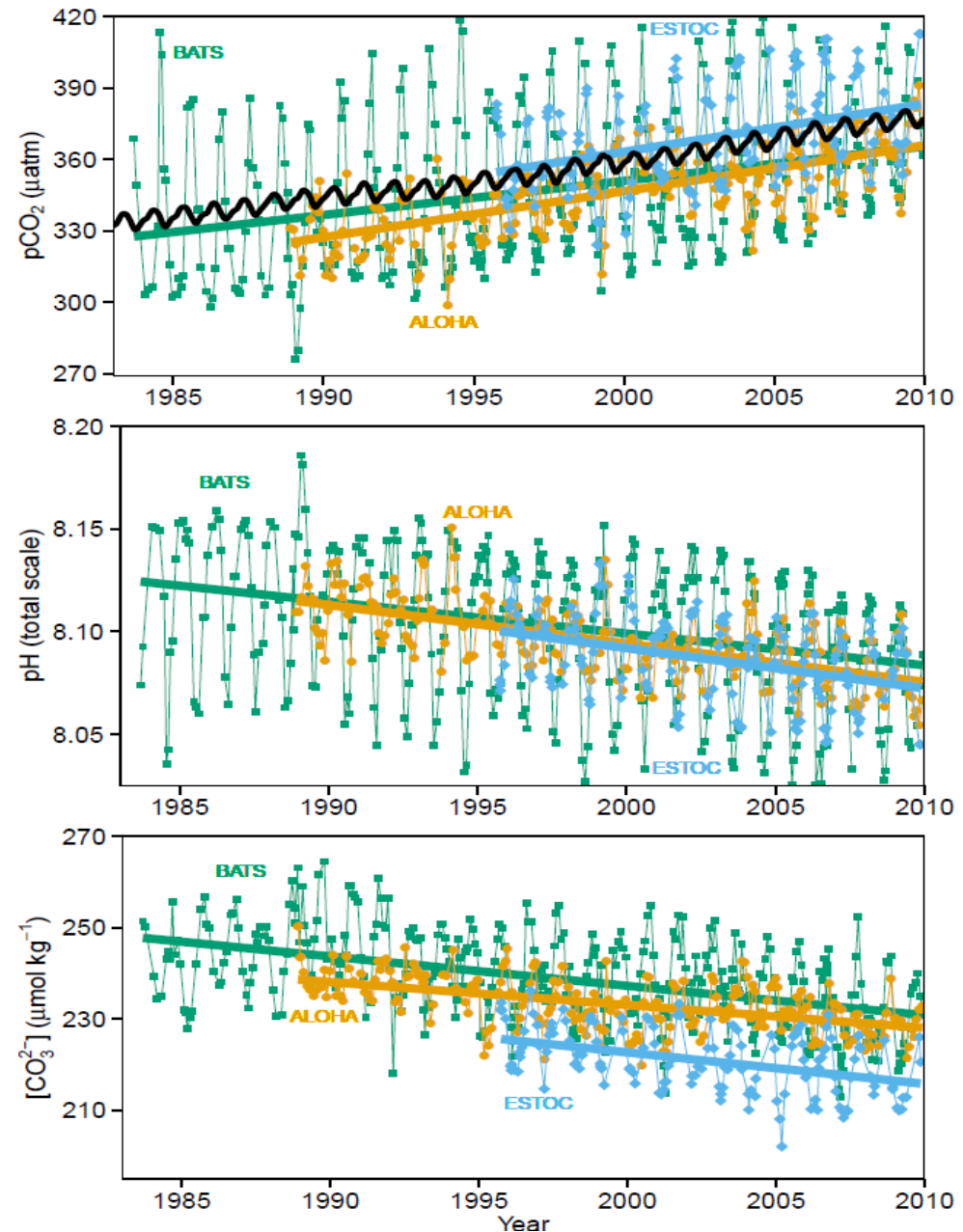




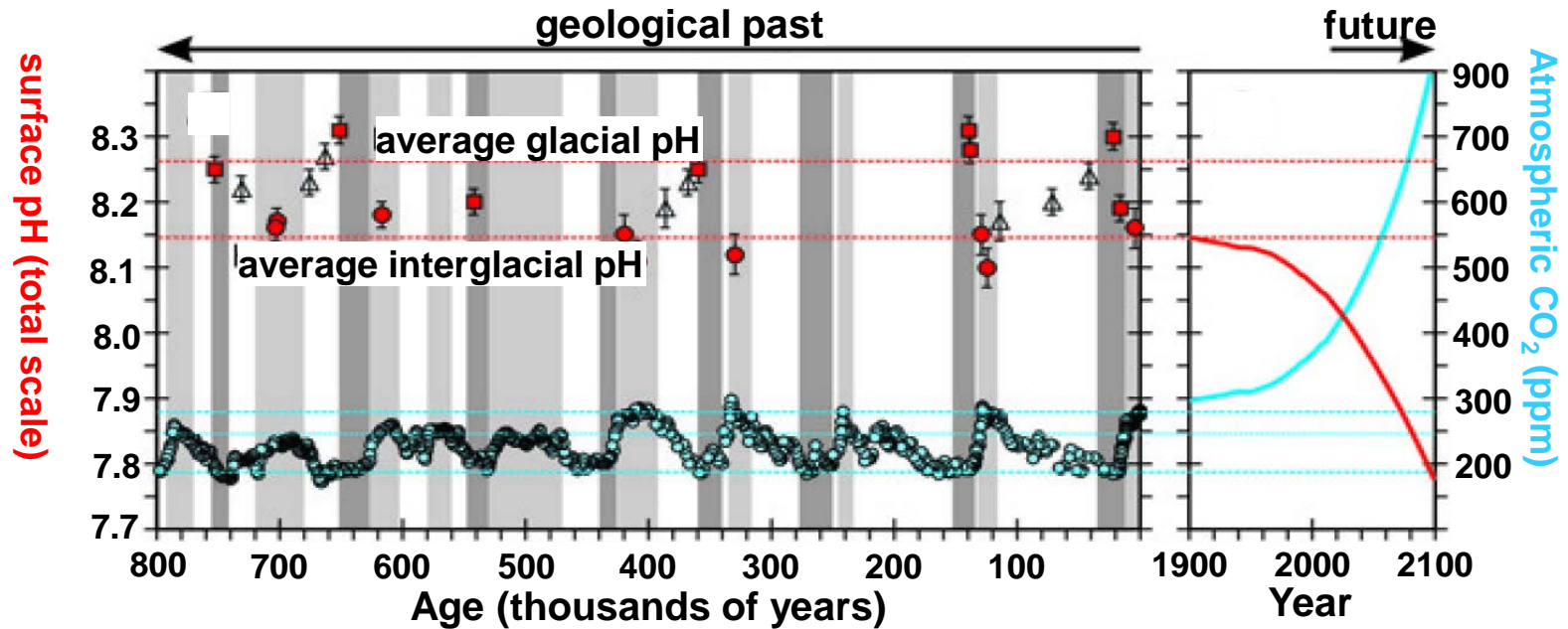
# Change in pH from ocean acidification already measurable



Data:  
*Bates (2007)*  
*Dore et al. (2009)*  
*Santana-Casiano et al. (2007)*  
*Gonzales-Dávila et al. (2010)*



# Today's rate of ocean acidification may be unprecedented

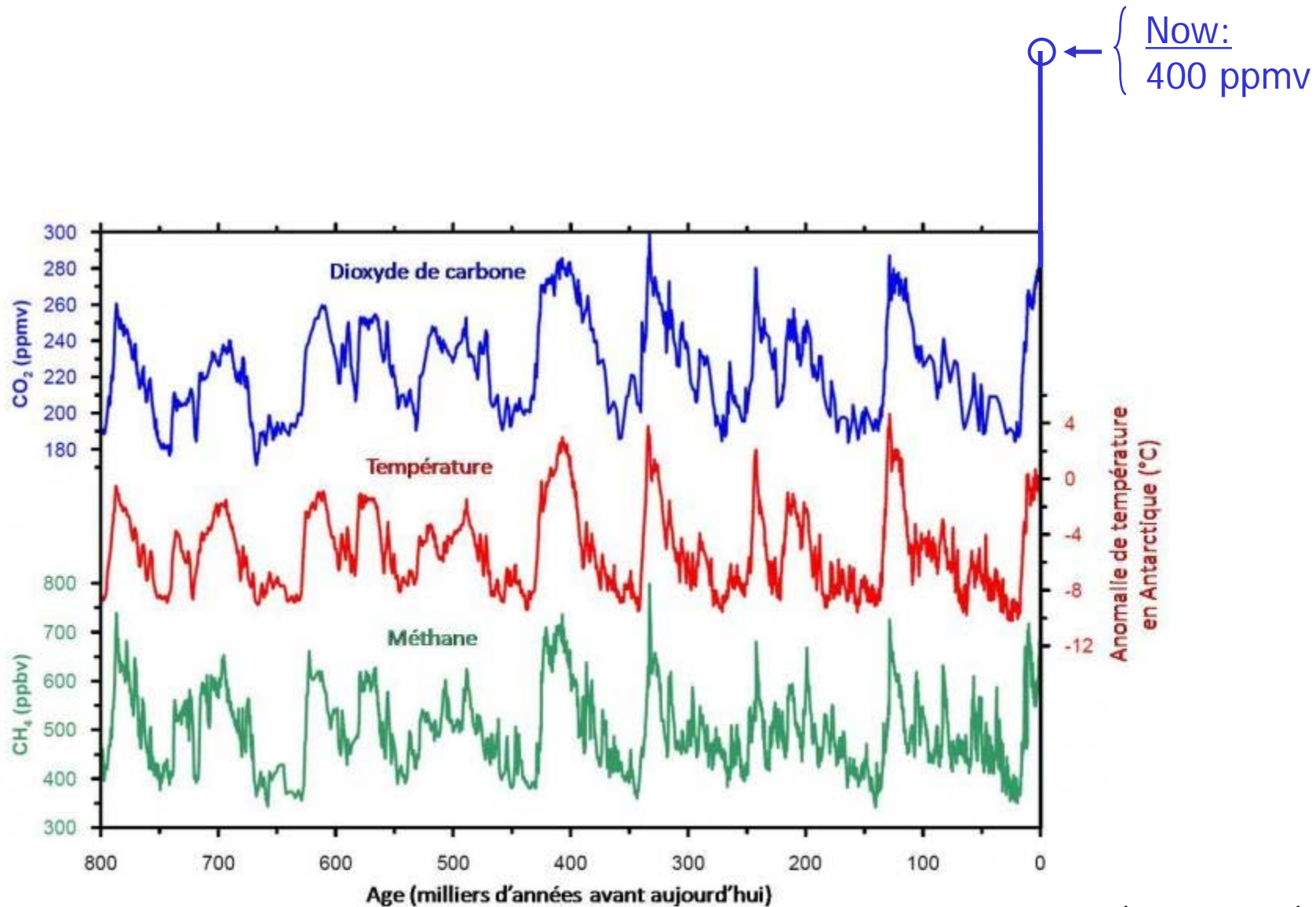


*Barker and Ridgwell (2012)*

## Current change:

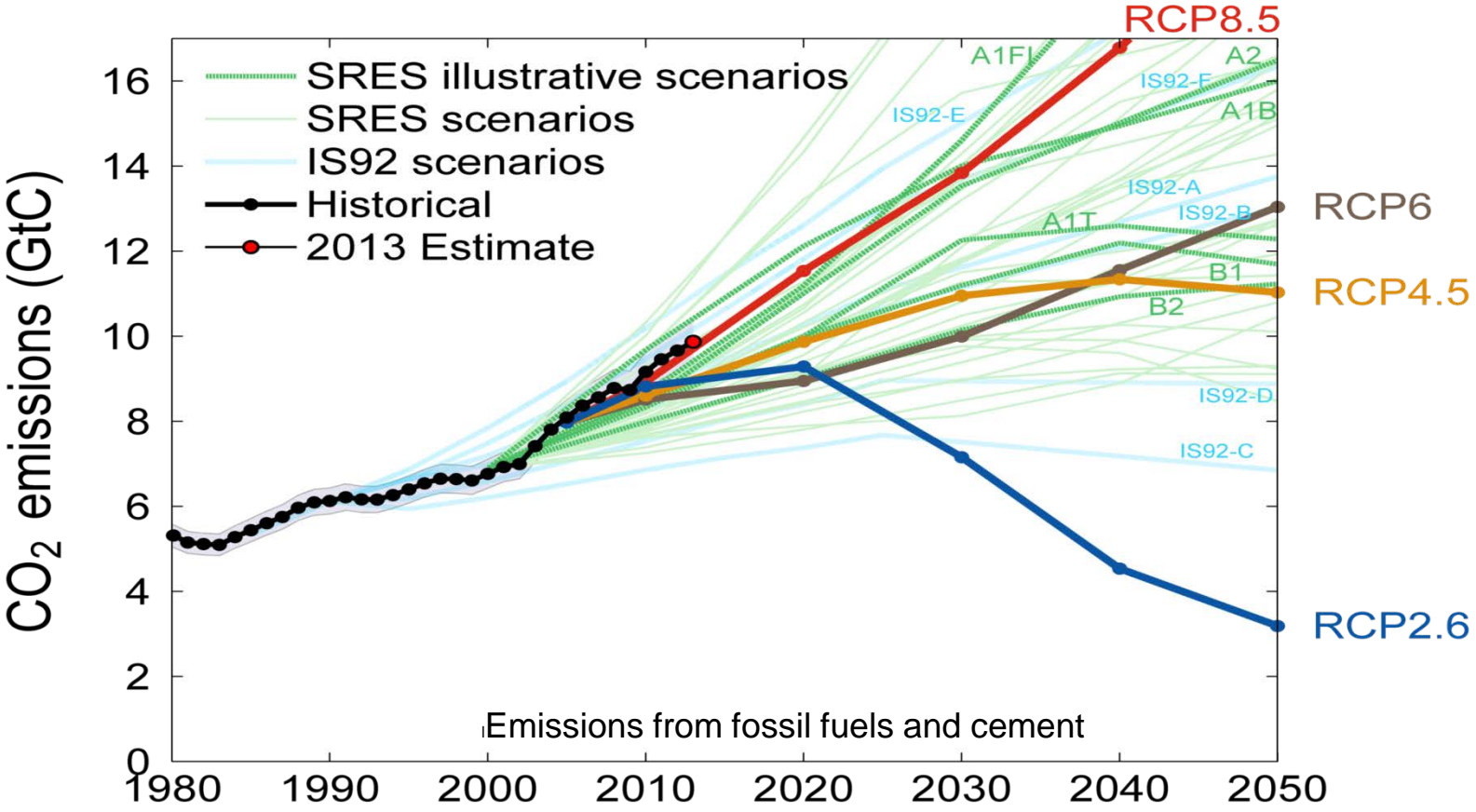
- overwhelms natural variations (last 800 000 years)
- may be 10 times faster than natural event (55 million years ago)
- rate may be unprecedented (over last 300 million of years)
- 26% increase in acidity ( $H^+$ ) during industrial era
- 100% increase (or more) projected by 2100

# Today's ocean acidification appears as instantaneous spike





# Current emissions tracking high emission scenarios

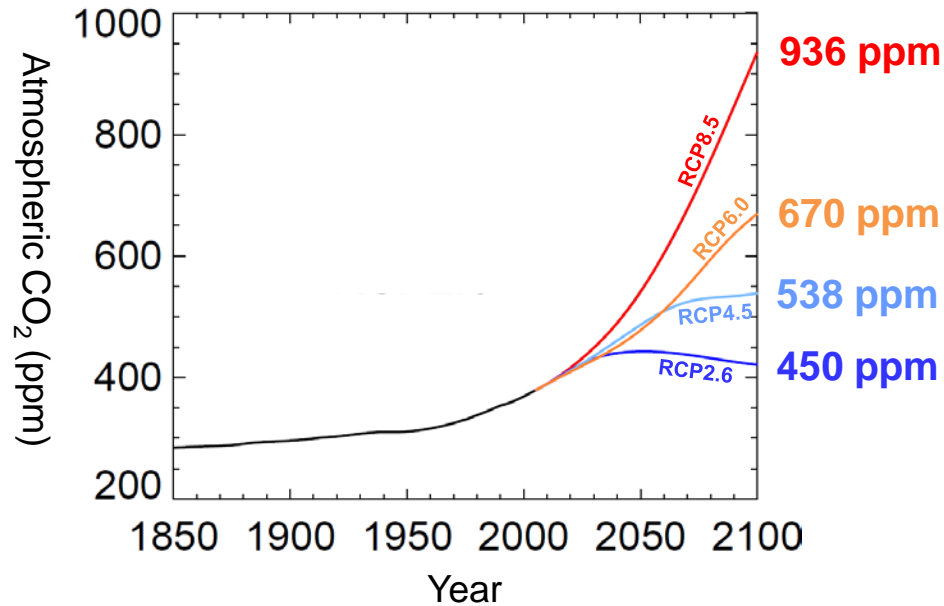


IPCC: 4 generations of emission scenarios

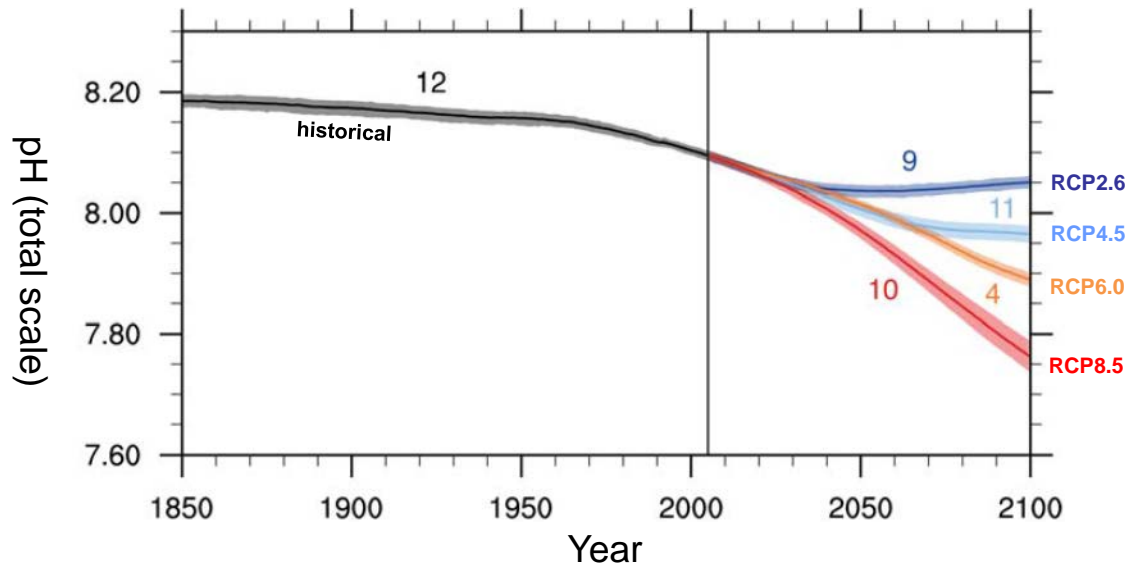
[Peters et al. 2012a](#); [CDIAC Data](#)

# Projected future surface pH ranges widely between scenarios

Future atmospheric CO<sub>2</sub> (latest IPCC scenarios)

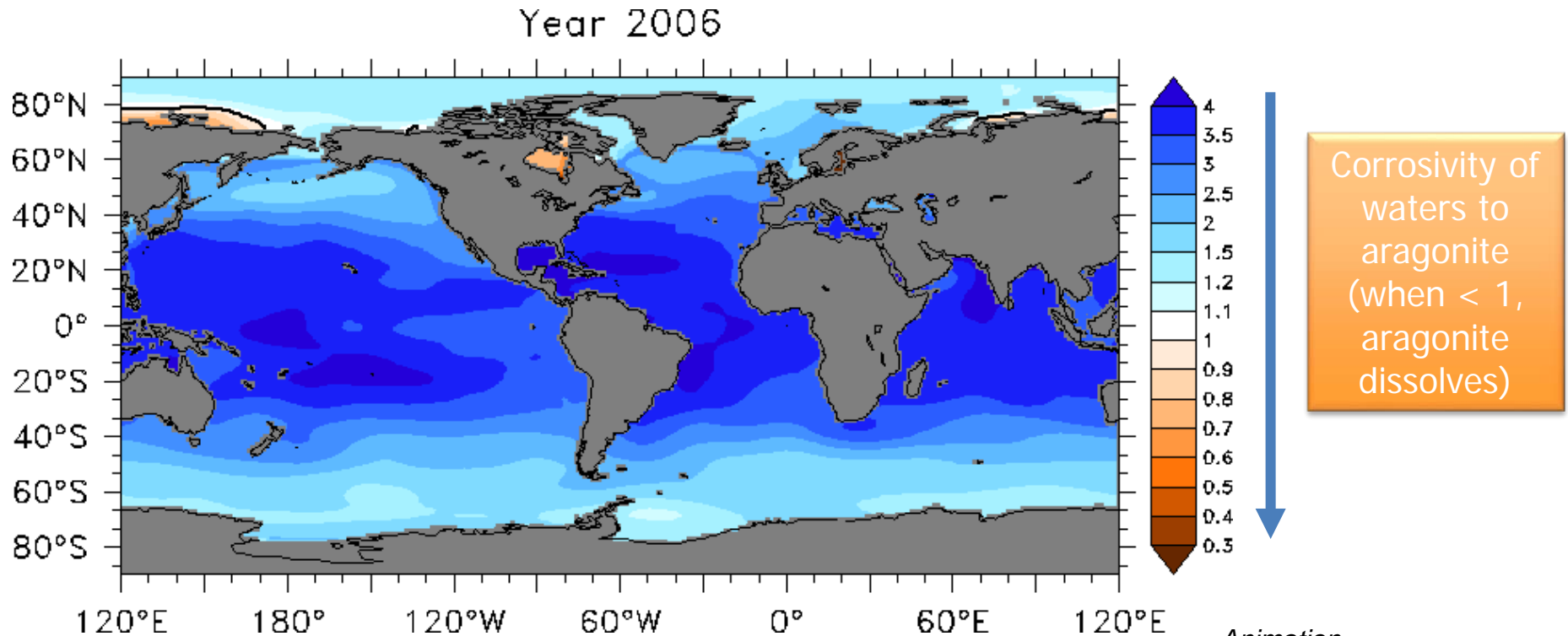


Intensity of ocean acidification (change in pH) varies by a factor of 3



# Polar oceans corrosive to shell material within decades

Models project that cold waters soon become corrosive to aragonite, a ( $\text{CaCO}_3$ ) mineral in some marine shells & skeletons



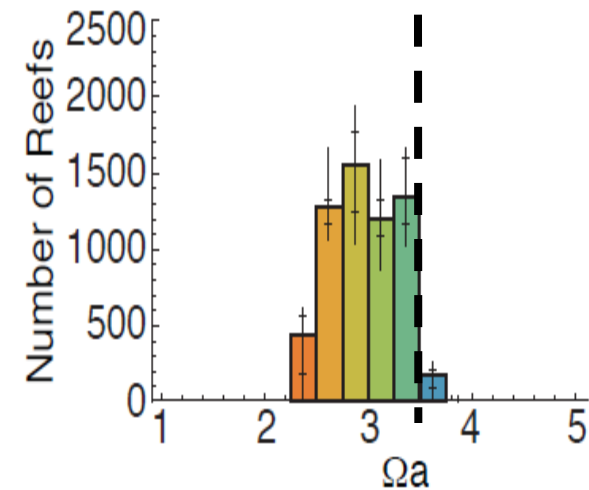
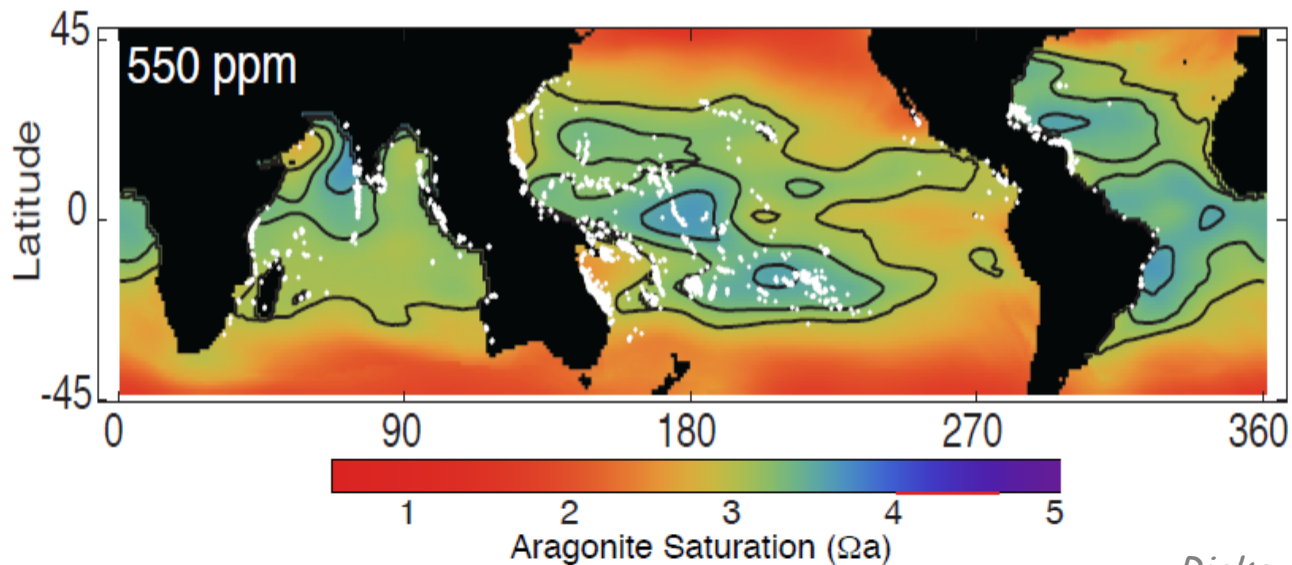
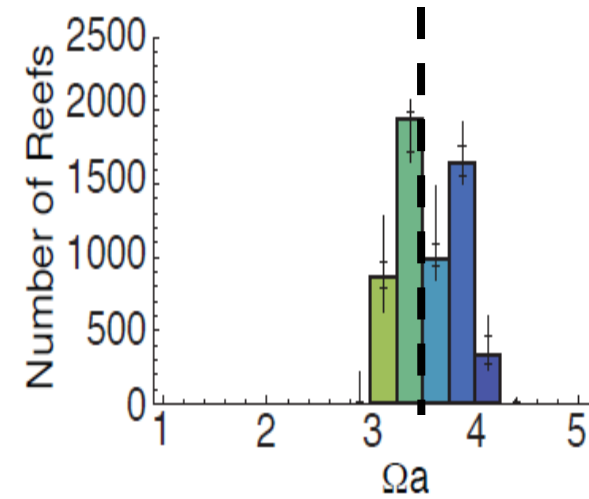
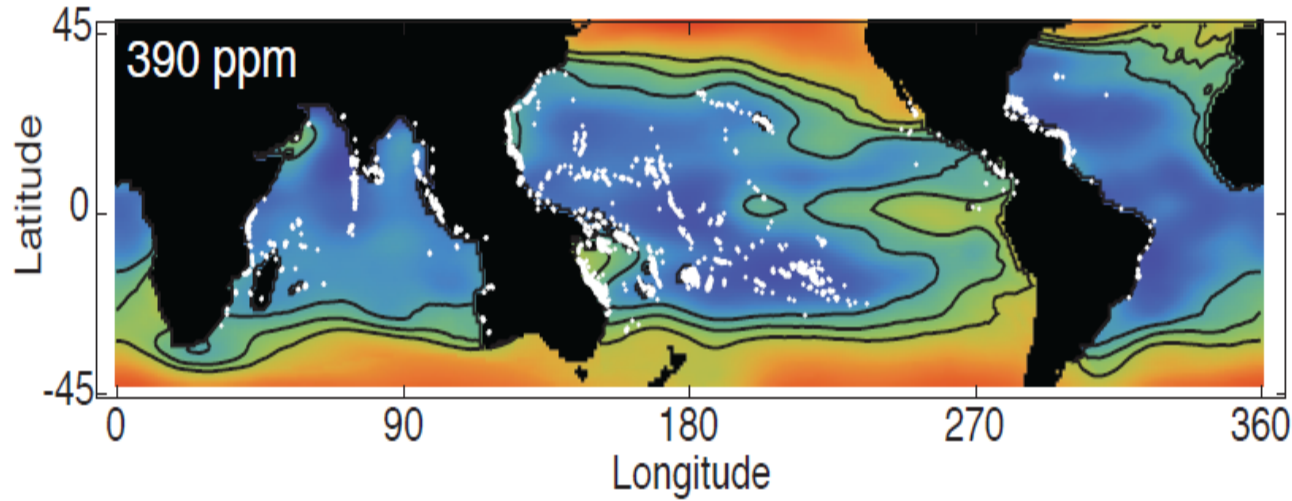
*Latest model projections (IPCC AR5 WG1, 2013)*

*Confirms original warnings: Orr et al. (2005), Caldeira & Wickett (2005), Steinacher et al. (2009)*

see also Bopp et al. (2013)

# Most tropical corals exposed to potentially unsustainable chemical conditions by mid-century (e.g., $\Omega_{\text{arag}} < 3.0$ )

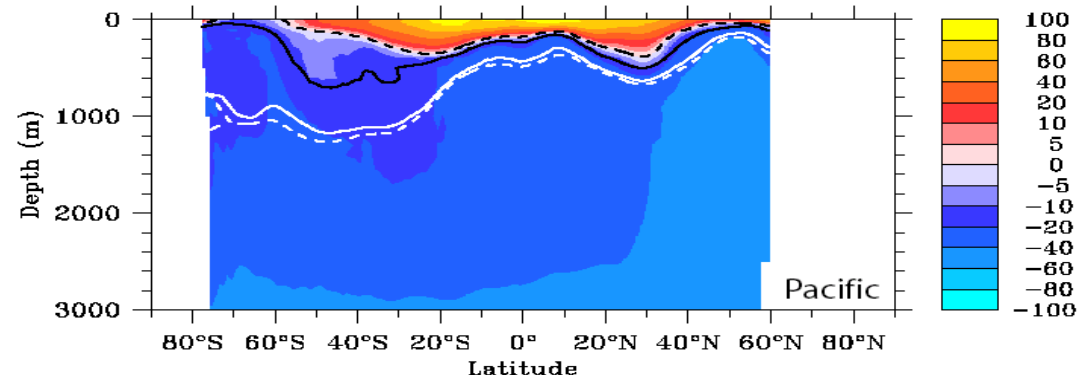
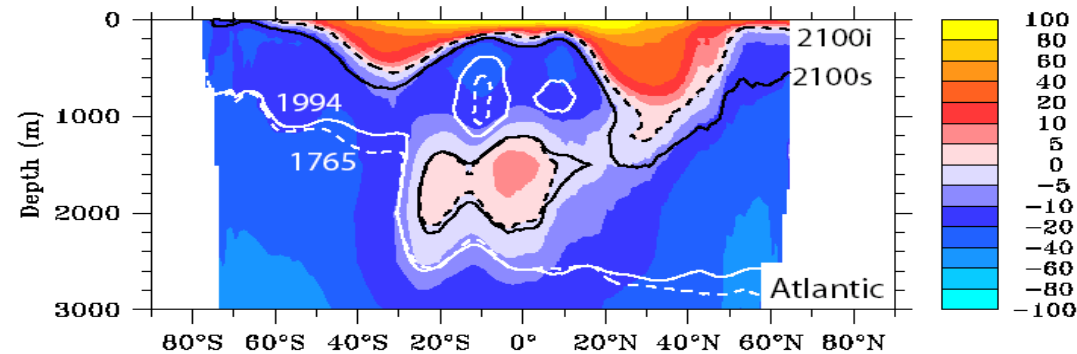
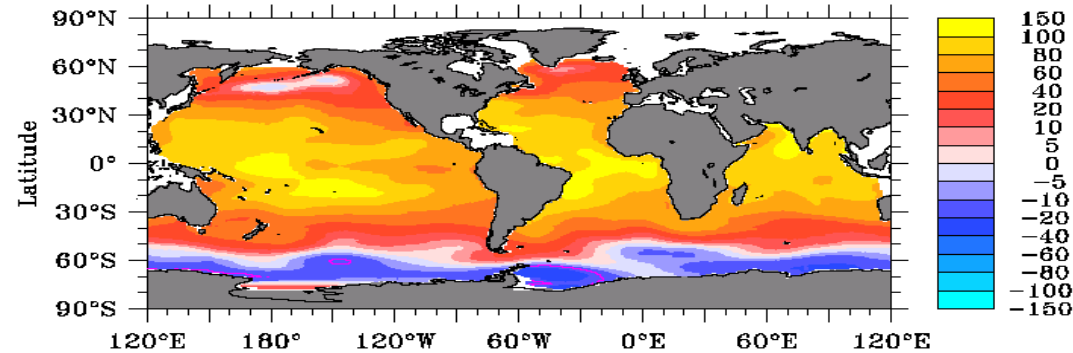
Analysis of 13 Earth System Models (CMIP5)



Ricke et al. 2013 (Env. Res. Letters)

# By 2100... Large changes in subsurface saturation state ( $\Delta[\text{CO}_3^{2-}]_A$ ) [in $\mu\text{mol kg}^{-1}$ ]

- Surface undersaturation ( $\Delta[\text{CO}_3^{2-}]_A < 0$ )
  - Southern Ocean
  - Subarctic Pacific
- Shoaling of the aragonite saturation horizon (i.e.,  $\Delta[\text{CO}_3^{2-}]_A = 0$ )
  - Southern Ocean (by ~1000 m)
  - North Atlantic (by ~3000 m)





# Most cold-water corals (made of aragonite) will be exposed to these corrosive conditions before 2100

## Deep, cold-water corals:

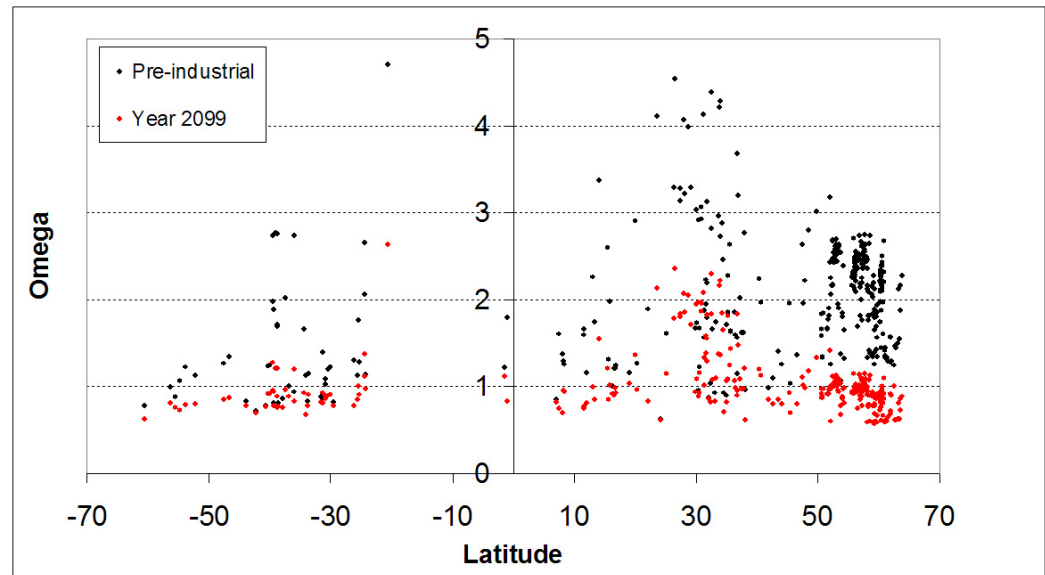
2005 : 95% avec  $\Omega_A > 1$

2100 : 35% avec  $\Omega_A > 1$

*Lophelia pertusa*



*L. pertusa* with expanded tentacles ready to capture zooplankton



*Guinotte et al. (2006)*  
*Davies et al. (2008)*  
*Fautin et al. (2009)*  
*Tittensor et al. (2010)*

# These corrosive conditions dissolve shells of sea butterflies

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*Movie: Brad Seibel, University of Rhode Island*

Sea butterfly shells ( $\text{CaCO}_3$ ) exposed to corrosive conditions expected by 2100



**Day 1**



**Day 2**



**Day 16**

Orr et al. (2005)

Fabry et al. (2008)

Comeau et al. (2009; 2011; 2012)

Lischka et al. (2011); Lischka & Riebesell (2012)

Bednarsek et al. (2012)

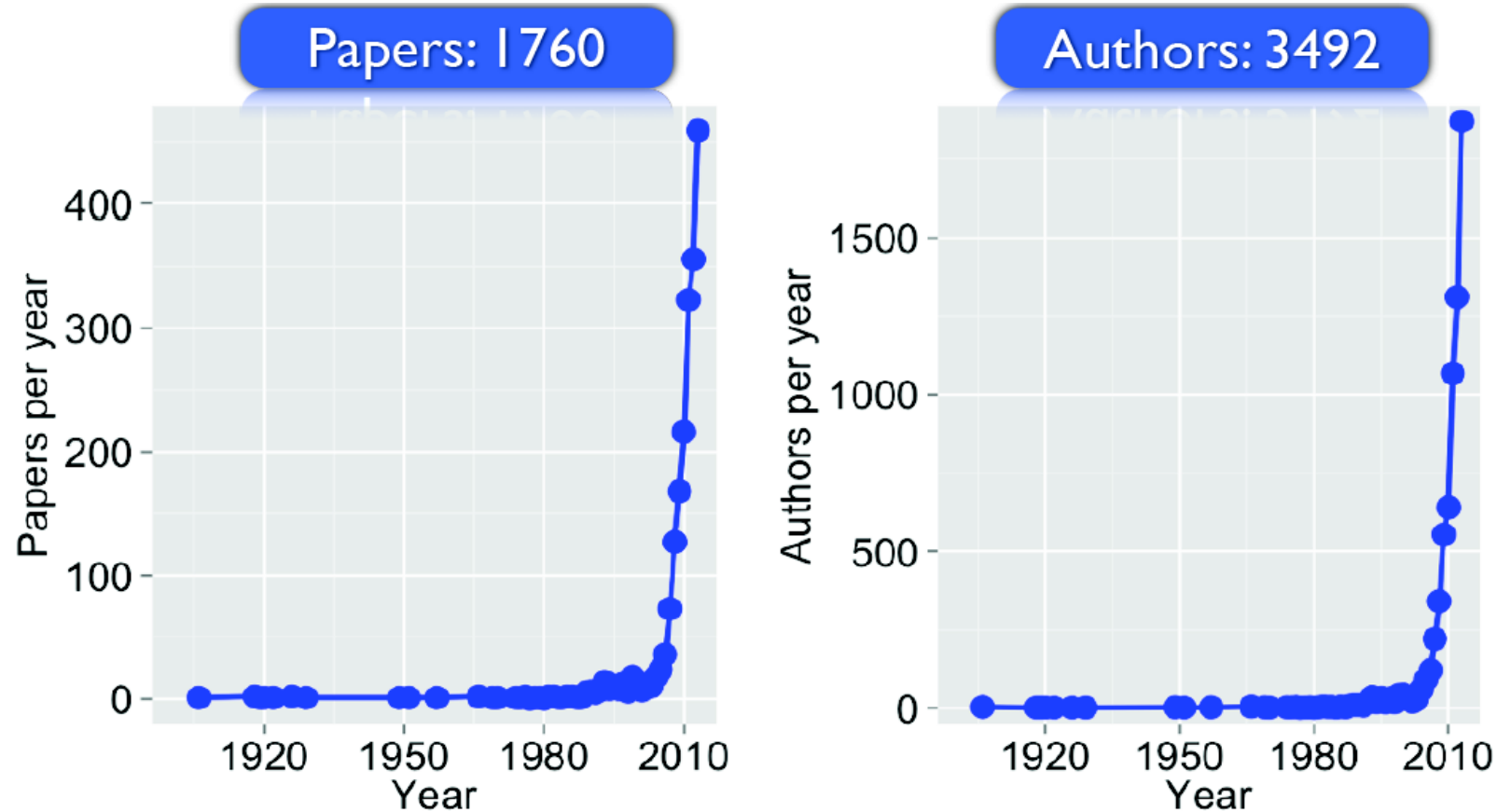
*Image: Victoria Fabry, California State University San Marcos*

# Manipulative studies used to evaluate biological responses

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- Lab perturbation experiments
- Field observations near CO<sub>2</sub> vents  
(natural, long-term perturbations)
- Mesocosm experiments (in the water; on the sediments)
- Free Ocean CO<sub>2</sub> Enrichment (FOCE) experiments

# Acidification research in its infancy; growing exponentially

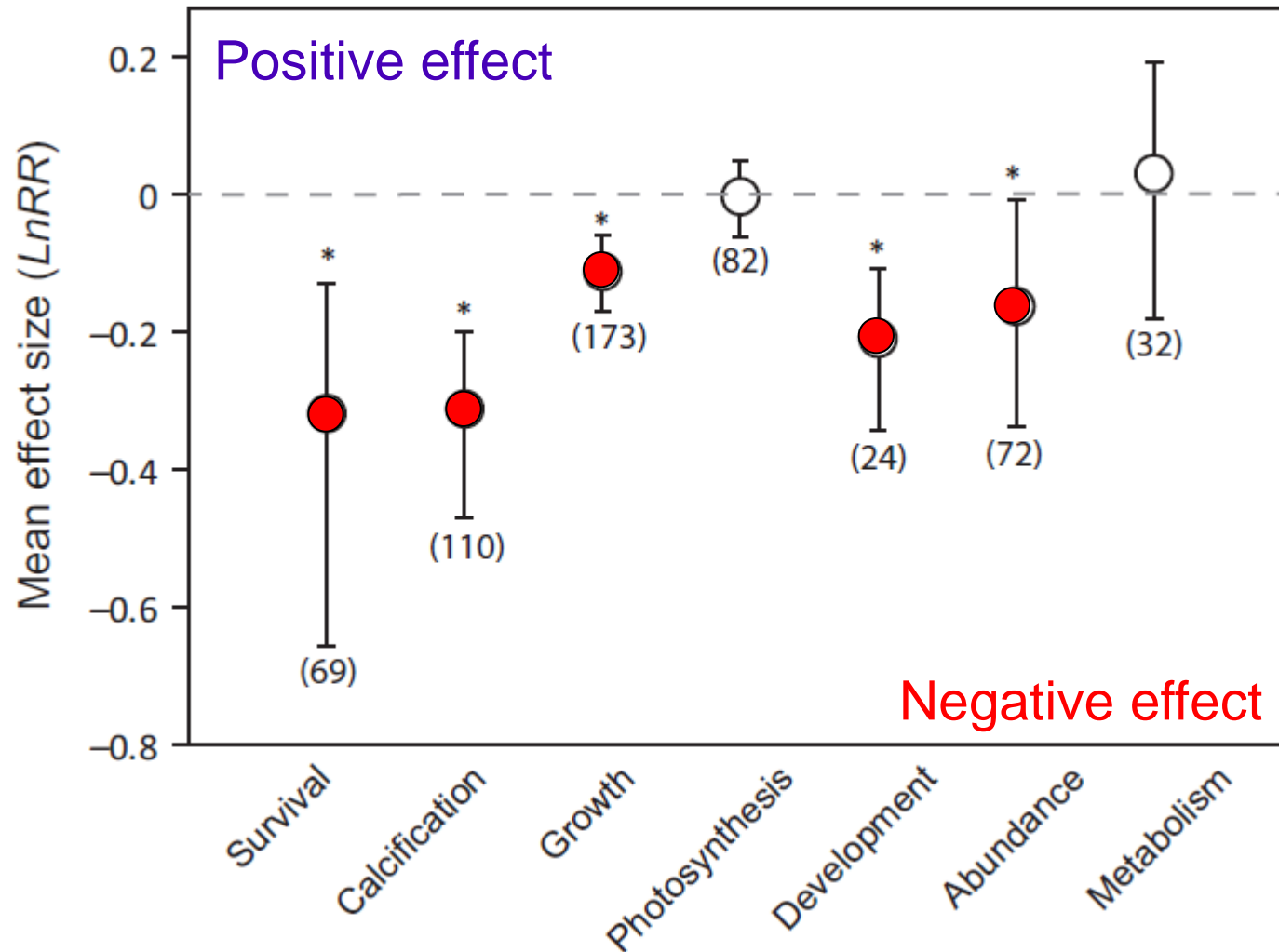


	2004	2005	2006	2007	2008	2009	2010	2011	2012	2013
<i>papers</i>	18	24	36	73	127	168	216	323	356	459
<i>authors</i>	58	92	120	221	341	554	640	1068	1312	1868

Update of Gattuso & Hansson (2011); Gattuso et al. (2011)

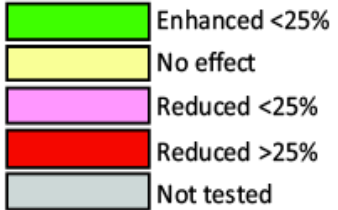
OA: #1 research front in Ecol. & Environ. Sci. (Thompson-Reuters Web of Knowledge)





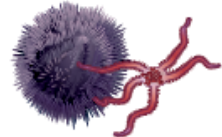
# Acidification adversely affects diverse groups










# But large diversity of effects on different groups



Taxa	Response	Mean Effect
 Calcifying algae	Survival	
	Calcification	
	Growth	
	Photosynthesis	-28%
	Abundance	-80%
 Corals	Survival	
	Calcification	-32%
	Growth	-23%
	Photosynthesis	
	Abundance	-47%
 Coccolithophores	Survival	
	Calcification	-9%
	Growth	
	Photosynthesis	
	Abundance	
 Molluscs	Survival	-34%
	Calcification	-40%
	Growth	-17%
	Development	-25%
	Abundance	
 Echinoderms	Survival	
	Calcification	
	Growth	-10%
	Development	-11%
	Abundance	

 Crustaceans	Survival	
	Calcification	
	Growth	
	Development	
	Abundance	
 Fish	Survival	
	Calcification	
	Growth	
	Development	
	Abundance	
 Fleshy algae	Survival	
	Calcification	
	Growth	+22%
	Photosynthesis	
	Abundance	
 Seagrasses	Survival	
	Calcification	
	Growth	
	Photosynthesis	
	Abundance	
 Diatoms	Survival	
	Calcification	
	Growth	+12%
	Photosynthesis	
	Abundance	

# Acidification likely to change marine ecosystems

Organisms react differently

## Synthesis of existing experimental studies

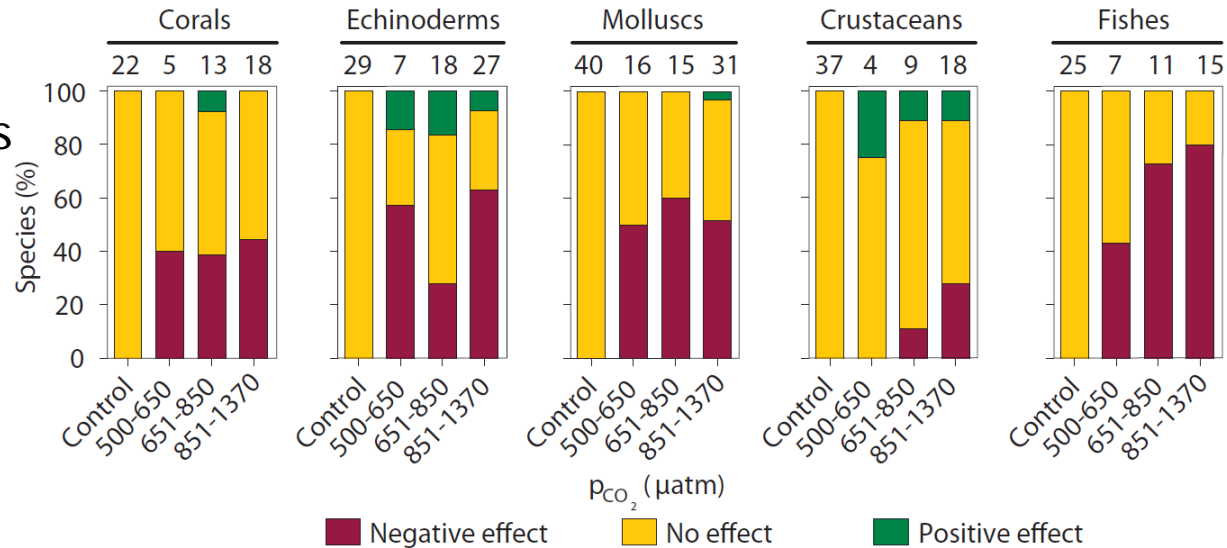
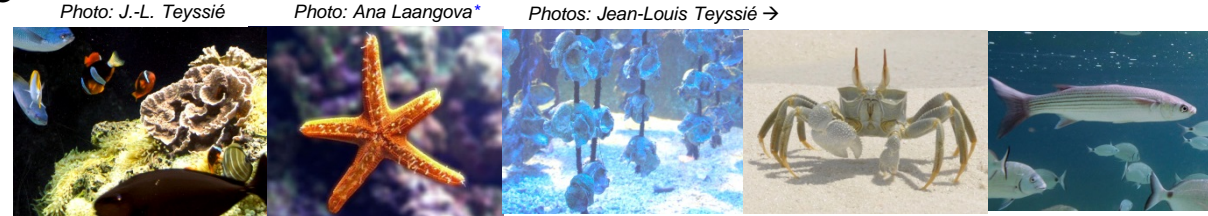
Corals and shell builders decline

Seagrasses may increase

Fish become disoriented

Predators affected by prey loss

Potential fish catch decline



Wittmann & Pörtner (2013, Nature Clim. Change)

# Ocean areas naturally rich in CO<sub>2</sub> confirm expected future trends

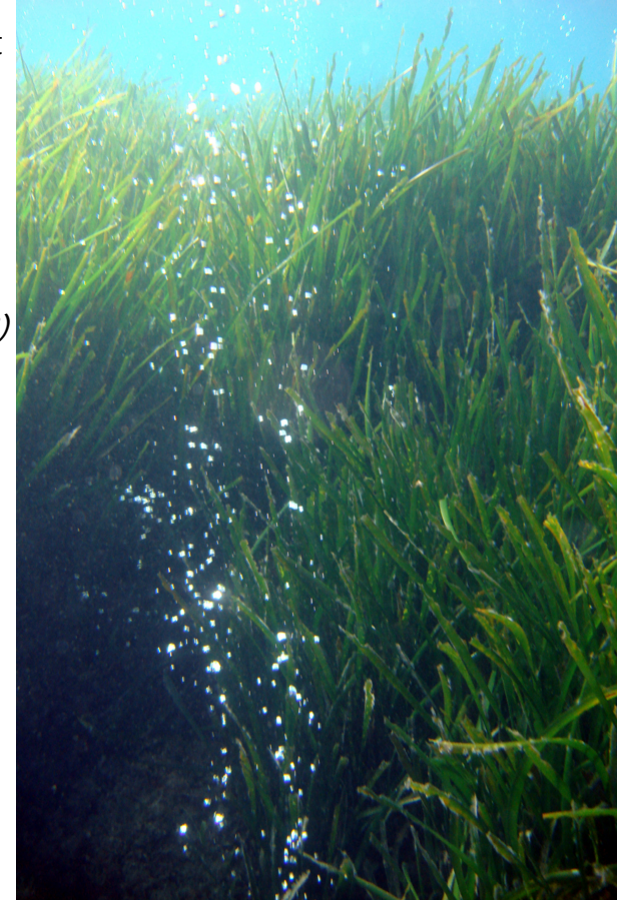
- Less biodiversity
- Fewer calcifiers
- More fragile shells
- More invasive species
- More seagrasses, degraded corals

CO<sub>2</sub> bubbles rise from seafloor at Ischia, Bay of Naples, a natural lab to study acidification

*Hall-Spencer et al. (2008)*

*Rodolfo-Metalpa et al. (2008)*

*Photo: Steve Ringman, Seattle Times*



*Photo: Jason Hall-Spencer,  
University of Plymouth*

Another natural CO<sub>2</sub> vent site in Papua, New Guinea, used to study effects of acidification on corals



# Ocean acidification will also affect humans

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- Fish is primary source of animal protein for 1 billion people, mostly in developing countries (FAO)
- Coral reefs provide
  - home for millions of species
  - storm protection for coastlines
  - income from tourism
  - biodiversity legacy for future
- Ocean acidification already affecting oyster industry (U.S. west coast)
- Ocean acidification may well affect aquaculture, fisheries, and human livelihoods
- Ocean acidification not happening in isolation



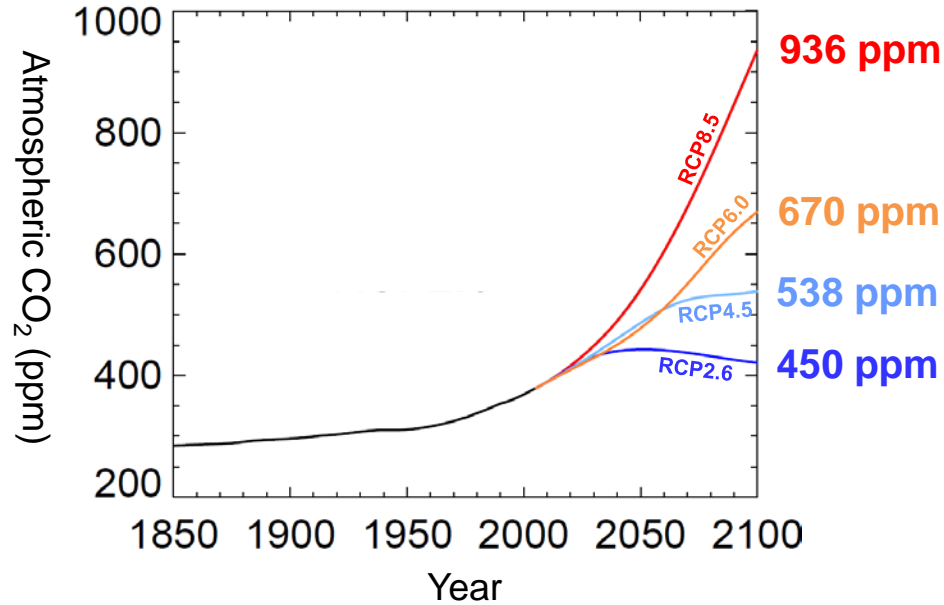
*Photo: Rodolfo Quevenco, IAEA*



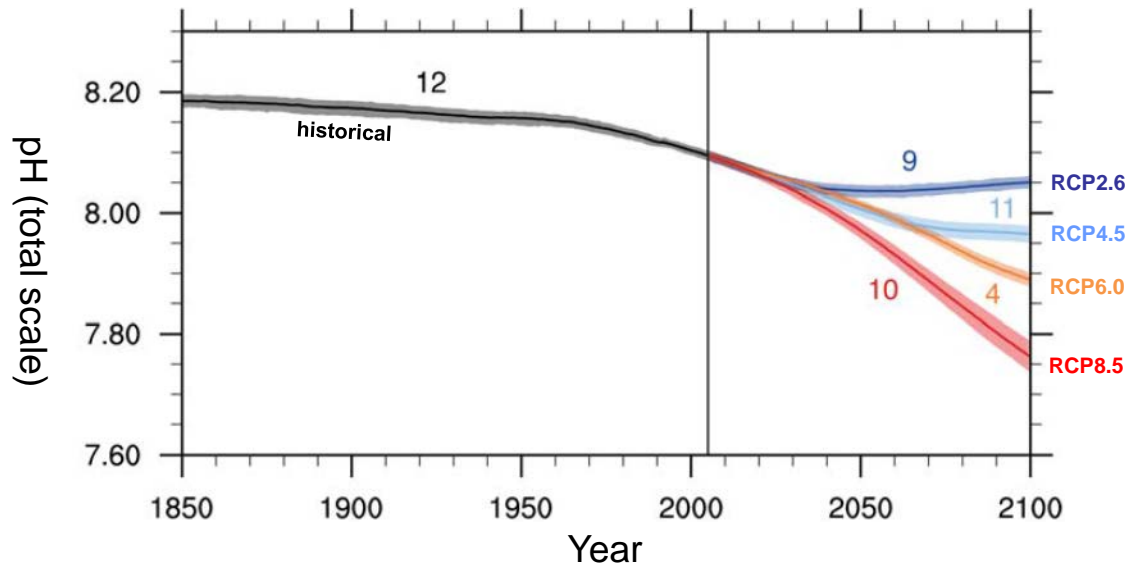
*Photo: Jean-Louis Teyssié, IAEA*

# The intensity of ocean acidification depends on us

Future atmospheric CO<sub>2</sub>  
(latest IPCC scenarios)

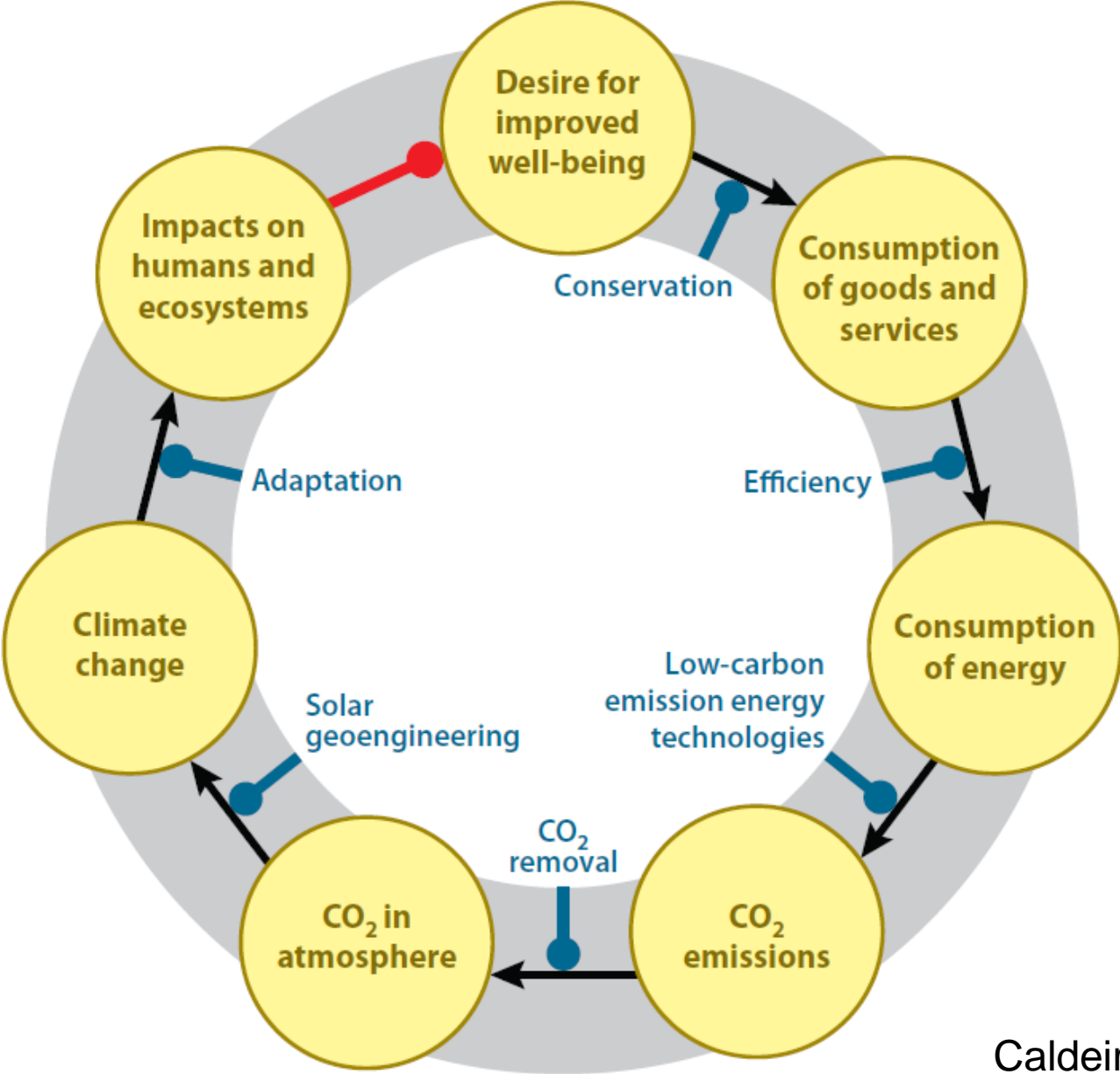


Intensity of ocean acidification (change in pH)  
varies by a factor of 3

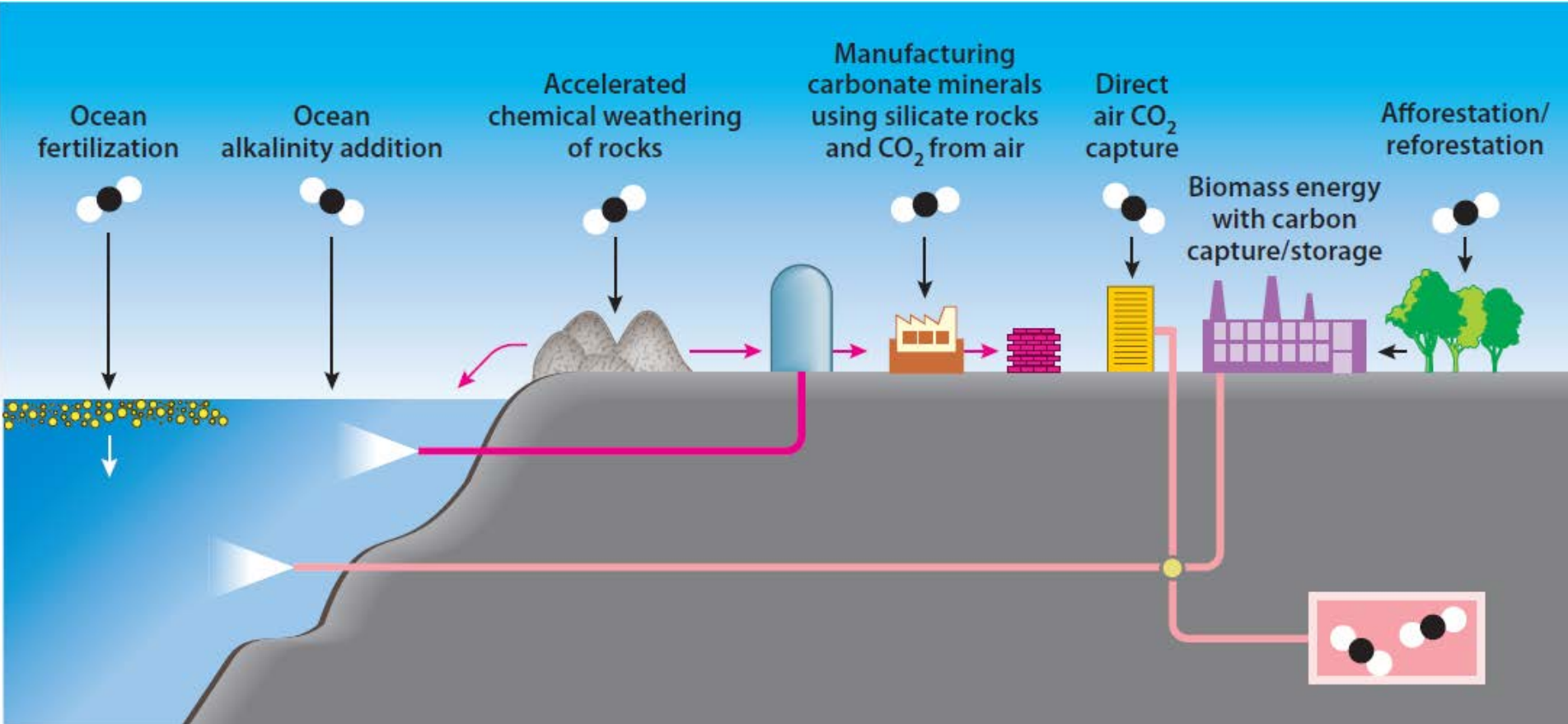




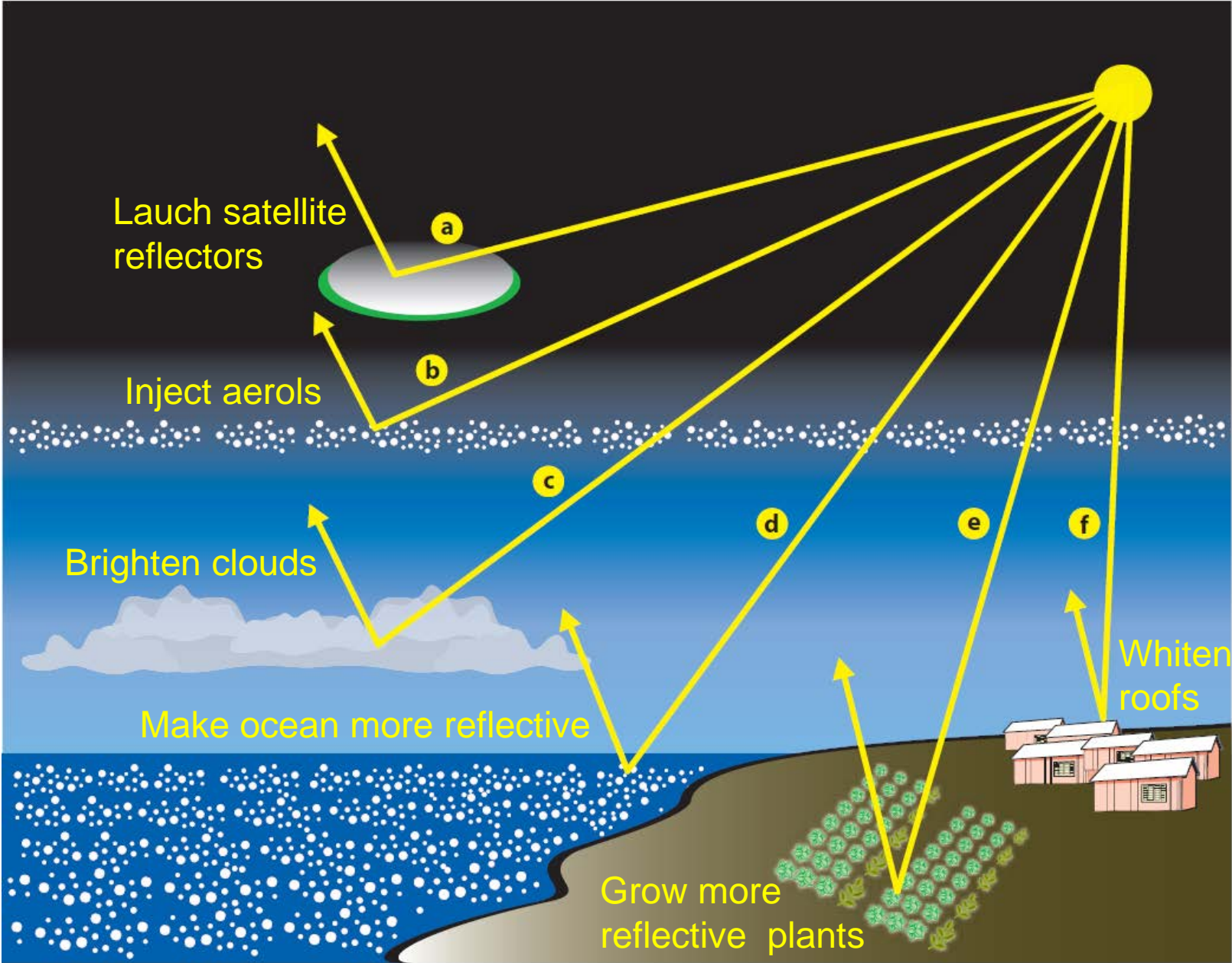
# Diverse measures required to limit CO<sub>2</sub> buildup



# Many proposals for CO<sub>2</sub> removal (CDR)

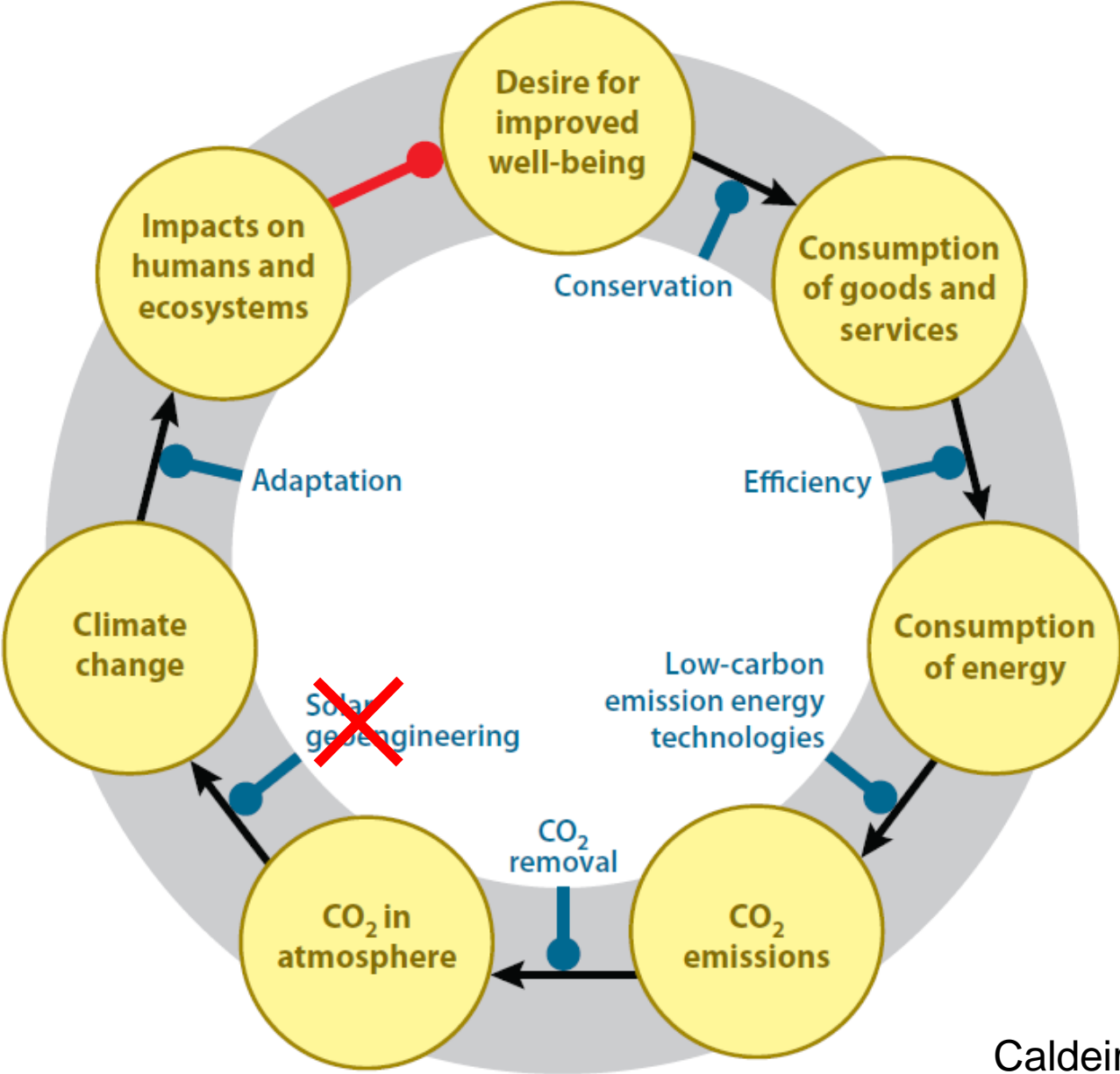


# Solar radiation management – no fix for acidification

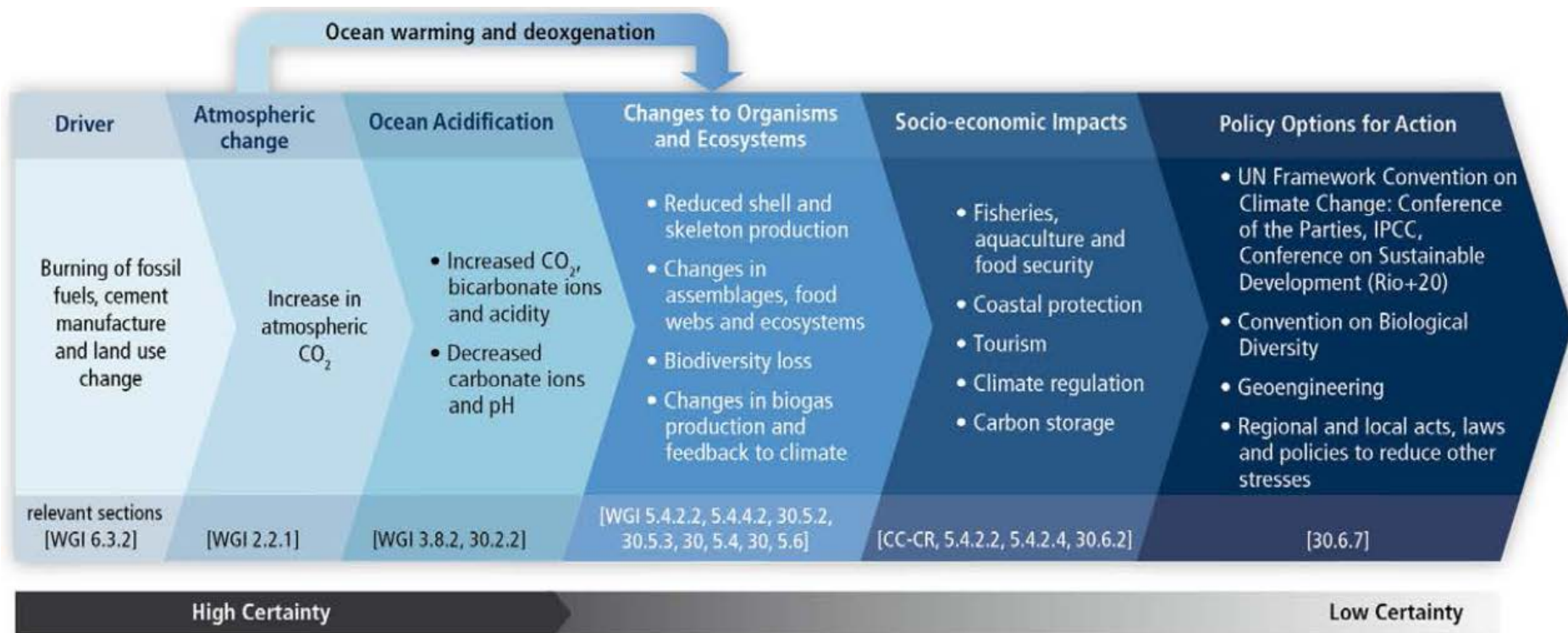


Caldeira et al. (2013)

# SRM would worsen ocean acidification



# Uncertainty grows as we try to assess how acidification affects us and how to manage it



Gattuso et al., Box CC-OA. Ocean Acidification, IPCC WGII, Final Draft, Chapter 6, 31 March 2014.



# Dissemination and outreach

Messages for Rio+20

Making it clear

A special introductory guide for policy advisers and decision makers

20 FACTS about Ocean Acidification

SECOND INTERNATIONAL SYMPOSIUM ON THE OCEAN IN A HIGH-CO<sub>2</sub> WORLD MONACO - OCTOBER 6-8 2012

Manoia Declaration

Documents for policy makers – some written by EPOCA's Reference User Group of stakeholders

World leading website and blog on ocean acidification

Dialogue with policy makers and media at climate change negotiations in Copenhagen, Cancun, Capetown and Warsaw

award winning films one by school children and another by professional film makers

TIPPING POINT

the other CO<sub>2</sub> problem

ocean acidification

JEAN-PIERRE GATTUSO AND LINA HANSSON

After C. Turley

# Useful resources

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OA-ICC web page:

<http://www.iaea.org/ocean-acidification>

Acidification news stream (blog):

<http://news-oceanacidification-icc.org>

General information:

<http://ocean-acidification.net>